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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/576,385	02/20/2007	Steven S. C. Chuang	089498.0496.US	1882
<sup>39905</sup> Joseph J. Crima	7590 12/22/201 ı <b>ldi</b>	0	EXAM	IINER
Roetzel & Andı	ess		ESSEX, STEPHAN J	
222 S. Main St. Akron, OH 443			ART UNIT	PAPER NUMBER
			1727	
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# Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)	
	10/576,385	CHUANG, STEVEN S. C.	
Office Action Summary	Examiner	Art Unit	
	STEPHAN ESSEX	1795	
The MAILING DATE of this communication a Period for Reply	ppears on the cover sheet w	th the correspondence address	-
A SHORTENED STATUTORY PERIOD FOR REF WHICHEVER IS LONGER, FROM THE MAILING  - Extensions of time may be available under the provisions of 37 CFR after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory perions a finite to reply within the set or extended period for reply will, by state the Any reply received by the Office later than three months after the main earned patent term adjustment. See 37 CFR 1.704(b).	DATE OF THIS COMMUNIO 1.136(a). In no event, however, may a red will apply and will expire SIX (6) MON oute, cause the application to become AE	CATION.  Epply be timely filed  THS from the mailing date of this communicat  ANDONED (35 U.S.C. § 133).	
Status			
1) ■ Responsive to communication(s) filed on <u>28</u> 2a) ■ This action is <b>FINAL</b> . 2b) ■ The sum of the sum	nis action is non-final. vance except for formal matt	•	is
Disposition of Claims			
4) ☑ Claim(s) 1-36,41 and 42 is/are pending in the 4a) Of the above claim(s) is/are withdress.  5) ☐ Claim(s) is/are allowed.  6) ☑ Claim(s) 1-36,41 and 42 is/are rejected.  7) ☐ Claim(s) is/are objected to.  8) ☐ Claim(s) are subject to restriction and	rawn from consideration.		
Application Papers			
9) The specification is objected to by the Exami 10) The drawing(s) filed on 19 April 2006 is/are:  Applicant may not request that any objection to the Replacement drawing sheet(s) including the correction.  The oath or declaration is objected to by the	a) accepted or b) object ne drawing(s) be held in abeyar ection is required if the drawing	ce. See 37 CFR 1.85(a). s) is objected to. See 37 CFR 1.121	, ,
Priority under 35 U.S.C. § 119			
12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of:  1. Certified copies of the priority docume 2. Certified copies of the priority docume 3. Copies of the certified copies of the priority docume application from the International Bure * See the attached detailed Office action for a li	ents have been received. ents have been received in A riority documents have been eau (PCT Rule 17.2(a)).	pplication No received in this National Stage	
Attachment(s)  1) Notice of References Cited (PTO-892) 2) D Notice of Draftsperson's Patent Drawing Review (PTO-948)	Paper No(s	ummary (PTO-413) s)/Mail Date	
3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date <u>3/3/2009, 6/30/2006</u> .	5)  Notice of I 6)  Other:	nformal Patent Application 	

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### **DETAILED ACTION**

#### Election/Restrictions

1. Applicant's election without traverse of Group I, claims 1-36, in the reply filed on October 28, 2010 is acknowledged. Claims 37-40 were cancelled. New claims 41 and 42 were added.

## Claim Rejections - 35 USC § 103

- 2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 3. The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
  - 1. Determining the scope and contents of the prior art.
  - 2. Ascertaining the differences between the prior art and the claims at issue.
  - 3. Resolving the level of ordinary skill in the pertinent art.
  - 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 4. Claims 1-3, 5-7, 9-11, 13-26, 28, 29, 31-35, 41 and 42 are rejected under 35 U.S.C. 103(a) as being unpatentable over Gur et al. (hereinafter "Gur") (U.S. Pat. No.

5,376,469) in view of Barnett (U.S. Pub. No. 2002/0098401A1) and Stark et al. (hereinafter "Stark") (U.S. Pub. No. 2004/0126298).

Regarding claims 1, 2, 6, 7, 9-11, 23-26 and 28, Gur teaches a high temperature fuel cell 10 that uses an oxygen ion conducting solid electrolyte 30 (solid-oxide electrolyte) in conjunction with separate temperature zones 14, 16 to optimize the direct electrochemical conversion of carbon fuels 12 (solid-state organic fuel) to electrical energy (direct-electrochemical-oxidation fuel cell) (see col. 2, lines 18-22). The solid electrolyte 30 has first and second electrolyte surfaces 32, 34 on which are deposited first (anode) and second (cathode) electrodes 36, 38, respectively (see col. 2, lines 37-41). Oxygen is provided to the carbon fuel 12 (direct electrochemical oxidation) by supplying an oxygen containing gas to the second electrode 38, the driving force for conduction of the oxygen ions through the solid electrolyte 30 being provided by the difference between the activities of oxygen in the fuel compartment (anode) and in the oxygen containing gas supplied to the second electrode 38 (formation of oxygen ions) (see col. 2, lines 64-68; col. 3, lines 1-3).

Gur is silent to an electrochemical-reduction catalyst and an electrochemical-oxidation catalyst.

Barnett teaches a solid oxide fuel cell for the direct electrochemical oxidation of hydrocarbons in solid oxide fuel cells wherein the anode may comprise NiO-YSZ (see paragraph 62). The cathode may comprise La<sub>0.8</sub>Sr<sub>0.2</sub>MnO<sub>3</sub> (lanthanum strontium manganese oxide) or LSCF (see paragraphs 83 and 107). It would have been obvious to one of ordinary skill in the art to have utilized the anode and cathode materials of

Barnett in the fuel cell of Gur because the selection of a known material, which is based upon its suitability for the intended use, is within the ambit of one of ordinary skill in the art. See *In re Leshin*, 125 USPQ 416 (CCPA 1960) (see MPEP § 2144.07).

Stark teaches that the yttria stabilized zirconia catalysts used in fuel cells may comprising additional catalytically active components such as nickel oxides, may further comprise rhenium or platinum (Re-NiO/YSZ) (see paragraphs 50 and 68). It would have been obvious to one of ordinary skill in the art at the time the invention was made to have provided the catalyst of Stark in the fuel cell of modified Gur because the selection of a known material, which is based upon its suitability for the intended use, is within the ambit of one of ordinary skill in the art. See *In re Leshin*, 125 USPQ 416 (CCPA 1960) (see MPEP § 2144.07).

Regarding claims 3 and 29, Gur teaches that the carbon fuel **12** (solid-state organic fuel) may be coal (see col. 6, lines 27-32).

Regarding claims 5 and 33, Gur teaches that the carbon fuel **12** is heated to a temperature that favors complete oxidation of the carbon fuel **12** to CO<sub>2</sub> (at least 50 mol % CO<sub>2</sub>) (see col. 2, lines 56-61).

Regarding claims 13, 14, 31, 32, 41 and 42, Gur teaches that suitable solid electrolytes **30** may be generated by doping oxides of Bi or Zr with rare each oxides such as  $Y_2O_3$  (see col. 5, lines 7-11).

Regarding claims 15 and 16, Gur teaches that the fuel cell **10** comprises fluidized bed reactor **50** (housing) including a bottom surface **52** having gas inlets **25**. In operation, fine particles of carbon fuel **12** are introduced (feed passage) into fuel

compartment **20** of fluidized bed reactor **50**. A stream of non-reactive gas is blown through gas inlets **25** to distribute the fine particles of carbon fuel **12** throughout the fuel compartment **20** (see col. 7, lines 61-66).

Regarding claims 17, 19, 21, 22 and 35, Gur, Barnett and Stark do not explicitly teach the NO<sub>x</sub> concentration of the anode product, the CO concentration in the cathode product, the electrical current density, or the fuel-conversion efficiency. However, it is the position of the examiner that these concentrations are inherent, given that the fuel cell of modified Gur is structurally and chemically identical to that of the claimed invention. A reference which is silent about a claimed invention's features is inherently anticipatory if the missing feature is necessarily present in that which is described in the reference. Inherency is not established by probabilities or possibilities. See *In re Robertson*, 49 USPQ2d 1949 (1999).

Regarding claims 18 and 20, Gur teaches that the fuel cell **10** is typically operated at temperatures no greater than 1100°C (maximum operating temperature of about 1200°C (see claims 11-13).

5. Claims 4 and 30 are rejected under 35 U.S.C. 103(a) as being unpatentable over Gur, Barnett and Stark as applied to claims 1-3, 5-7, 9-11, 13-26, 28, 29, 31-35, 41 and 42 above, and further in view of Coors (U.S. Pat. No. 7,332,237) and Paisley (U.S. Pat. No. 6,680,137).

Regarding claims 4 and 30, Gur, Barnett and Stark are silent to peat, rice hulls and corn husks.

Coors teaches that fuel cells operating on solid carbon fuels may be pyrolized biomass. It would have been obvious to one of ordinary skill in the art at the time the invention was made to have utilized biomass fuel in the fuel cell of modified Gur because Coors teaches biomass as a functional equivalent of coal (see col. 13, lines 39-59).

Paisley teaches that biomass fuels include peat and agricultural waste (see col. 6, lines 29-32). It would have been obvious to one of ordinary skill in the art at the time the invention was made to have utilized peat or agricultural waste as the biomass fuel in the fuel cell of modified Gur because the selection of a known material, which is based upon its suitability for the intended use, is within the ambit of one of ordinary skill in the art. See *In re Leshin*, 125 USPQ 416 (CCPA 1960) (see MPEP § 2144.07).

6. Claims 8 and 27 are rejected under 35 U.S.C. 103(a) as being unpatentable over Gur, Barnett and Stark as applied to claims 1-3, 5-7, 9-11, 13-26, 28, 29, 31-35, 41 and 42 above, and further in view of Visco et al. (hereinafter "Visco") (U.S. Pub. No. 2006/0057295A1).

Regarding claim 8, Gur, Barnett and Stark are silent to a material having the formula A<sub>x</sub>B<sub>y</sub>CO<sub>3</sub>.

Visco teaches Sm<sub>0.5</sub>Sr<sub>0.5</sub>CO<sub>3</sub> as a functional equivalent of lanthanum strontium manganese oxide (see paragraph 54). It would have been obvious to one of ordinary skill in the art at the time the invention was made to have substituted Sm<sub>0.5</sub>Sr<sub>0.5</sub>CO<sub>3</sub> for the lanthanum strontium manganese oxide in the cathode of modified Gur because the

selection of a known material, which is based upon its suitability for the intended use, is within the ambit of one of ordinary skill in the art. See *In re Leshin*, 125 USPQ 416 (CCPA 1960) (see MPEP § 2144.07).

7. Claims 12 and 36 are rejected under 35 U.S.C. 103(a) as being unpatentable over Gur, Barnett and Stark as applied to claims 1-3, 5-7, 9-11, 13-26, 28, 29, 31-35, 41 and 42 above, and further in view of Giaquinta et al. (hereinafter "Giaquinta") (U.S. Pub. No. 2007/0010396A1).

Regarding claim 12, Gur, Barnett and Stark are silent to Cu oxide-Pt.

Giaquinta teaches an oxidation catalyst for fuel cells consisting essentially of platinum and copper oxides (see paragraphs 37 and 41). It would have been obvious to one of ordinary skill in the art at the time the invention was made to have provided the oxidation catalyst of Giaquinta in the fuel cell of modified Gur because the selection of a known material, which is based upon its suitability for the intended use, is within the ambit of one of ordinary skill in the art. See *In re Leshin*, 125 USPQ 416 (CCPA 1960) (see MPEP § 2144.07).

## Conclusion

8. Any inquiry concerning this communication or earlier communications from the examiner should be directed to STEPHAN ESSEX whose telephone number is (571) 270-7866. The examiner can normally be reached on Monday - Friday, 7:30-5:00 EST.

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Dah-Wei Yuan can be reached on (571) 272-1295. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

SJE

/Dah-Wei D. Yuan/ Supervisory Patent Examiner, Art Unit 1727